

Calmagite - EGTA. Colorimetric

Quantitative determination of magnesium





PACKAGING

Ref.: 101-0284 Cont.: 2 x 100 mL

Store at 2-8° C

CLINICAL SIGNIFICANCE

Magnesium is the second more abundant intracellular cation of the human body after potassium, being essential in great number of enzymatic and metabolic processes. Is a cofactor of all the enzymatic reactions that involve the ATP and comprises of the membrane that maintains the electrical excitability of the muscular and nervous cells. A low magnesium level is found in malabsortion syndrome, diuretic or aminoglucoside therapy; hyperparathyroidism or diabetic acidosis. Elevated concentration of magnesium is found in uremia, chronic renal failure, glomerulonephritis, Addisons's disease or intensive anti acid therapy 1,4,5 .

Clinical diagnosis should not be made on a single test result; it should integrate clinical and other laboratory data.

PRINCIPLE OF THE METHOD

Magnesium form a purple coloured complex when reacts with calmagite in alkaline solution (Note 1).

The intensity of the color formed is proportional to the magnesium concentration in the sample¹.

REAGENTS

R 1	Amino-methyl-propanol	1 mmol/L
Buffer	EGTA	0.21 mmol/L
R 2 Chromogen	Calmagite	0.30 mmol/L
MAGNESIUM	Magnesium aqueous primary standard 2 mg/dL	
CAL		

Optional (not included in the kit)

Contro-N	Ref.: 101-0252	4 x 5 mL	Lyophilized human	
Collifo-IN	Ref.: 101-0083	20 x 5 mL	control serum	
Contro-P	Ref.: 101-0253	4 x 5 mL	Lyophilized human	
	Ref.: 101-0084	20 x 5 mL	control serum	

PRECAUTIONS

R1/RT:Corrosive (C):R35:Causes severe burns.

PREPARATION

Working reagent (WR):

Mix equal volumes of R 1 Buffer and R 2 Chromogen.

The working reagent is stable for 4 days at refrigerator (2 - 8° C) or 24 h at room temperature (15 - 25° C).

STORAGE AND STABILITY

All the components of the kit are stable until the expiration date on the label when stored tightly closed at 2 - 8° C protected from light and contaminations prevented during their use.

Do not use reagents over the expiration date.

Signs of reagent deterioration:

- Presence of particles and turbidity.
- Blank absorbance (A) at 520 nm \ge 1.4.

ADDITIONAL EQUIPMENT

- Spectrophotometer or colorimeter measuring at 520 nm.
- Matched cuvettes 1.0 cm light path.
- General laboratory equipment (Note 2)

SAMPLES

- Serum, heparinized plasma¹:

Free of hemolysis and separated from cells as rapidly as possible.

Do not use oxalates or EDTA as anticoagulant.

Stability: 7 days at 2 - 8° C.

- Urine1:

Should be acidified to pH 1 with HCl.

If urine is cloudy; warm the specimen to 60° C for 10 min. to dissolve precipitates.

Dilute the sample 1/10 with distilled water and multiply the result by 10. Stability: 3 days at 2 - 8° C

PROCEDURE

Notes: CHRONOLAB SYSTEMS has instruction sheets for several automatic analyzers. Instructions for many of them are available on request.

MAGNESIUM CAL: Proceed carefully with this product because due its nature it can get contamined easily.

Interference by calcium is prevented by the use of EGTA¹.

It is recommended use disposable material to avoid calcium or magnesium contamination. If glassware is used the material should be scrupulously clean with H2SO4 - K2Cr2O7 and then thoroughly rinsed with distilled water and dried before use.

Calibration with the aqueous standard may cause a systematic error in automatic procedures. It is recommended to use a serum Calibrator. Use clean disposable pipette tips for its dispensation.

- Assay conditions:
 - Cuvette: 1 cm light path
- Adjust the instrument to zero with distilled water.
- Pipette into a cuvette:

	Blank	Standard	Sample
WR (mL)	1.0	1.0	1.0
Standard ^(Note 3-4) (µL)		10	
Sample (µL)			10

- Mix and incubate for 5 min at room temperature or 3 min a 37° C.
- Read the absorbance (A) of the samples and calibrator, against the Blank. The colour is stable for at least 30 minutes.

CALCULATIONS

(A) Sample – (A) Blank x 2 (Standard conc.) = mg/dL magnesium de in (A) Standard -(A) Blank

the sample

Conversion factors:

 $mg/dL \times 0.412 = mmol/L \text{ or}$

 $0.5 \text{ mmol/L} = 1.0 \text{ mEq/L} = 1.22 \text{ mg/dL} = 12.2 \text{ mg/L}^{1}$.

QUALITY CONTROL

Control sera are recommended to monitor the performance of assay procedures.

If control values are found outside the defined range, check the instrument, reagents and calibrator for problems.

Each laboratory should establish its own Quality Control scheme and corrective actions if controls do not meet the acceptable tolerances.

REFERENCE VALUES¹

Serum or plasma:

 $1.6 - 2.5 \text{ mg/dL} \cong 0.66 - 1.03 \text{ mmol/L}$

Urine:

$$24 - 244 \text{ mg}/24 \text{ h} \cong 2 - 21 \text{ mEq/L/24 h}$$

These values are for orientation purpose; each laboratory should establish its own reference range.

PICC048e	CHRONOLAB SYSTEMS, S.L., C/Diputación 260, 08007 Barcelona, SPAIN
V 2015/3	Tel. +34 617722466, www.chronolab.com, e-mail: comercial@chronolab.es





Magnesium

Calmagite - EGTA. Colorimetric

Quantitative determination of magnesium

PERFORMANCE CHARACTERISTICS

Measuring range: From *detection limit* of 0.2 mg/dL to *linearity limit* of 5 mg/dL. If the results obtained were greater than linearity limit, dilute the sample 1/2 with NaCl (9 g/L) and multiply the result by 2.

Precision:

	Intra-assay (n=20)	
Mean (mg/dL)	2.39	4.01
SD	0.02	0.07
CV (%)	1.18	1.73

Inter-assay (n=20)	
2.27	4.14
0.07	0.13
2.99	3.22

Sensitivity: 1 mg/dL = 0.055 A.

Accuracy: Results obtained using CHRONOLAB reagents (y) did not show systematic differences when compared with other commercial reagents (x).

The results obtained using 50 samples were the following:

Correlation coefficient (r): 0.998

Regression equation: y=0.971x + 0.145

The results of the performance characteristics depend on the analyzer used.

INTERFERENCES

Hemolysis and anticoagulants other than heparin¹.

A list of drugs and other interfering substances with magnesium determination has been reported by Young et. al².

BIBLIOGRAPHY

- Farrell E C. Magnesium. Kaplan A et al. Clin Chem The C.V. Mosby Co. St Louis. Toronto. Princeton 1984; 1065-1069.
- Young DS. Effects of drugs on Clinical Lab. Tests, 4th ed AACC Press, 1995
- Young DS. Effects of disease on Clinical Lab. Tests, 4th ed AACC 2001.
- Burtis A et al. Tietz Textbook of Clinical Chemistry, 3rd ed AACC 1999
- Tietz N W et al. Clinical Guide to Laboratory Tests, 3rd ed AACC 1995.